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Enantioselective Construction of Spiro{cyclopropane-1,4'-bicyclo[3.3.0]oct-1-en-3-ones}¹

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Abstract: Intramolecular Pauson-Khand reactions of 1,6-enynes 3a-c with a methylenecyclopropane terminator and a chiral acetal moiety adjacent to the triple bond gave spiro{cyclopropane-1,4'-bicyclo[3.3.0]oct-1-en-3-ones} 5a-c in good yields with a diastereoselectivity of up to 6.4:1. The major diastereomer of 5b was converted to enantiomerically pure bicyclo[3.3.0]octane-3,8-dione 8, which showed a negative peak at 287 nm in the CD curve, consistent with an assumed (5R) configuration.

Intramolecular Pauson-Khand reactions $(PKR)^2$ of 1,6-enynes³ to form bicyclo[3.3.0]oct-1-en-3-ones proceed particularly well when the double bond is part of a methylenecyclopropane moiety.⁴ In order to demonstrate the full potential of this approach to spiro{cyclopropane-1,4'-bicyclo[3.3.0]oct-1-en-3-ones} we have tested the possibility of asymmetrically inducing the cyclization with a chiral auxiliary adjacent to the triple bond in the 1,6-enyne, especially since only a few examples of PKR with asymmetric induction have been reported to date.⁵



As Magnus et al. have demonstrated,³ high diastereoselectivities can be obtained in the intramolecular PKR. On the basis of their mechanistic rationalization it was conceived that a cyclopropylidenealkyne of type 3 with a C_2 -symmetric acetal moiety next to the triple bond – by favouring an intermediate with the configuration as in 4 – might lead to an asymmetric induction in the cyclization step.⁶ The 6-cyclopropylidene-1-hexyn-3-one 2 was prepared by reacting the ester 1⁷ with lithium trimethylsilylacetylide in the presence of boron trifluoride etherate⁸ (80% yield), and converted to the acetals 3a-c by transacetalization⁹ with the appropriate ethanediol¹⁰ and trimethyl orthoformate (Scheme 1).

Although heavily substituted, the trimethylsilyl protected alkynes $3a-c^{12}$ underwent trialkylamine oxide promoted PKR¹³ quite well (63-76% yield). The diastereoselection in the cyclization of 3a with a dimethylsubstituted acetal moiety was only 2:1, according to GC, but increased to 5:1 for the diphenyl 3b and even 6.4:1 for the dicyclohexyl derivative $3c.^{11}$

The diastereomers of **5b** and **5c** could be separated by column chromatography.¹⁴ Further transformations were performed with the major diastereomer of **5b**, which was assumed to have (5R) configuration on the basis of the adopted mechanism with a preferred intermediate **4**.¹⁵ In order to determine the absolute configuration of the bicyclo[3.3.0] octenone skeleton, it was attempted to cleave off the chiral auxiliary in the acetal moiety. Upon treatment of **5b** with acetone and *p*-toluenesulfonic acid, however, not only acetal

Scheme 2



cleavage, but also double bond migration occurred to yield the achiral bicyclo[3.3.0]octenedione 6 (Scheme 2). Therefore, lithium dimethylcuprate was first added to the enone moiety in 5b, before the dioxolane was hydrolyzed. Surprisingly, acetal cleavage in the resulting 7 (2-endo/2-exo = 1 : 7) was quite slow, within 2 h at room temperature only protiodesilylation occurred to give 9. Only upon prolonged heating under reflux, 9

eventually gave the dione 8 with $[\alpha]_D^{20} = -148$ ° (c = 1, CHCl₃). The CD curve with a negative peak at 287 nm (ellipticity of 550°) is consistent with the absolute configuration¹⁶ being (5*R*) as assumed on the basis of the preferred orientation of bulky groups in the intermediate 4.

This model study demonstrates a new asymmetric variant of the intramolecular Pauson-Khand reaction. It is conceivable that further tailoring of the acetal moiety may lead to even higher diastereoselectivities. To achieve this, it is however essential to have the methylenecyclopropyl terminator in the precursor enyne; control experiments showed that enynes without this end group reacted only poorly and only under more drastic conditions (hexane, 110 °C), albeit with nearly the same diastereoselectivity.

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- 11. Cf. Hoffmann, R. W.; Ditrich, K.; Köster, G.; Stürmer, R. Chem. Ber. 1989, 122, 1783-1789. We are grateful to Professor R. W. Hoffmann for a generous gift of (R,R)-1,2-dicyclohexylethanediol.
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- All new compounds (except 3a, 3b, 5a, 6, and the minor diastereomers of 5b,c) were fully characterized by spectroscopic techniques (¹H NMR, ¹³C NMR, IR, MS) and their molecular formulas established by elemental analysis and/or HRMS. Physical data for relevant compounds: 2: IR (neat) 3030, 2970, 2930, 2160 (C=C), 1680 (C=O), 1255, 1115, 855, 770 cm⁻¹; ¹H NMR (250 MHz, CDCl₃) δ 0.32 (s, 9 H), 0.98–1.06 (m, 4 H), 2.49–2.59 (m, 2 H), 2.68–2.77 (m, 2 H), 5.72–5.82 (m, 1 H); ¹³C NMR (62.9 MHz, CDCl₃) δ -0.83, 1.96, 2.05, 26.15, 44.59, 97.70, 101.91, 115.67, 122.57 3a: IR (neat) 2940, 2900, 2830, 1420, 1365, 1240, 1190, 1070, 855, 830, 750 cm^{-1:} ¹H NMR (250 MHz, C₆C₆) δ 0.14 (s, 9 H), 0.82–0.95 (m, 4 H), 1.02 (d, 3 H), 1.18 (d, 3 H), 2.32–2.41 (m, 2 H), 2.86–2.90 (m, 2 H), 3.47 (dq, i H), 4.08 (dq, i H), 5.57–5.84 (m, 1 H); ¹³C NMR (62.9 MHz, CDCl₃) δ -0.23, 1.99, 16.44, 17.84, 26.22, 39.55, 79.22, 79.43, 80.08, 101.82, 104.61, 117.23, 121.30. 3b: IR (KBr) 3052, 2960, 2903, 1253, 1216, 1194, 1122, 1095, 1060, 1041, 1026, 976, 951, 843, 768, 701, 533

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cm⁻¹; ¹H NMR (250 MHz, CDCl₃) δ 0.14 (s, 9 H), 0.95–0.96 (m, 4 H), 2.14–2.21 (m, 2 H), 2.45–2.54 (m, 2 H), 4.58 (d, 1 H), 4.99 (d, 1 H), 5.75–5.81 (m, 1 H), 7.12–7.29 (m, 10 H); ¹³C NMR (62.9 MHz, CDCl₃) δ -0.24, 2.03, 26.13, 39.40, 86.20, 86.51, 89.55, 103.55, 104.24, 117.14, 121.58, 126.72, 127.12, 128.15, 128.34, 128.45, 135.70, 137.59. - 3c: IR (neat) 3050, 2925, 2853, 2166 (C=C), 1450, 1250, 863, 843 cm⁻¹; ¹H NMR (250 MHz, C₆D₆) δ 0.14 (s, 9 H), 0.85 (s, 4 H), 1.00-2.08 (m, 21 H), 2.30-2.43 (m, 3 H), 2.26-2.38 (m, 2 H), 3.54-3.63 (m, 1 H), 4.01-4.09 (m, 1 H), 5.88-5.96 (m, 1 H); ¹³C NMR (62.9 MHz, C₆D₆) δ -0.17, 2.11, 2.27, 26.43, 26.70, 26.90, 27.00, 28.12, 29.64, 30.21, 30.58, 40.62, 41.58, 41.74, 84.29, 84.55, 87.72, 102.46, 107.11, 117.91, 121.49. - 5a: IR (neat) 2980, 2880, 1680 (C=O), 1600, 1235, 1110, 1070, 840; ¹H NMR (250 MHz, C₆D₆) δ 0.48–0.46 (m, 1 H), 0.51, 0.52 (s, 9 H), 0.57–0.70 (m, 1 H), 0.90–1.12 (m, 7 H), 1.30–1.54 (m, 2 H), 1.86-2.12 (m, 7 H), 2.30-2.36 (m, 1 H), 3.42-3.56 (m, 1 H), 3.60-3.75 (m, 1 H); ¹³C NMR (62.9 MHz, C₆D₆) δ 0.21, 0.33, 14.41, 14.50, 14.61, 15.64, 16.59, 18.72, 25.21, 25.61, 33.46, 33.83, 40.86, 40.89, 78.20, 78.64, 79.00, 79.52, 110.49, 111.15, 138.55, 139.00, 188.67, 189.00, 211.42. - 5b IR (KBr) 2937, 1695 (C=O), 1121, 845, 699, 668 cm⁻¹; ¹H NMR (250 MHz, C₆D₆) δ 0.60 (s, 9 H), 0.55–0.66 (m, 2 H), 0.71–0.79 (m, 2 H), 1.41–1.54 (m, 2 H), 2.10–2.22 (m, 1 H), 2.32– 2.42 (m, 1 H), 3.08 (dd, 1 H), 4.92 (d, 1 H), 5.21 (d, 1 H), 7.10-7.27 (m, 8 H), 7.37-7.42 (m, 2 H); ¹³C NMR (62.9 MHz, C_6D_6) δ 0.45, 14.43, 15.05, 24.65, 34.14, 39.93, 50.12, 84.85, 86.27, 113.34, 126.68, 127.29, 128.49, 128.87, 136.41, 137.85, 138.15, 187.15, 211.48; [α]² = 218.1°; minor diastereomer: ¹H NMR (250 MHz, C_6D_6) δ 0.63 (s. 9 H), 0.64–0.77 (m, 2 H), 1.01–1.08 (m, 1 H), 1.14–1.22 (m, 1 H), 1.36–1.55 (m, 2 H), 2.21–2.27 (m, 2 H), 2.87 (dd, 1 H), 5.07 (d, 1 H), 5.27 (d, 1 H), 7.11–7.45 (m, 1 H). – ¹³C NMR (62.9 MHz, C₆D₆) & 0.27, 14.43, 14.9025.69, 33.39, 41.55, 52.85, 83.91, 85.07, 111.67, 135.72, 138.88, 140.36, 187.70, 211.47. - 5c: IR (neat) 2927, 2852, 1694 (C=O), 1244, 1123, 844 cm⁻¹; ¹H NMR (250 MHz, C₆D₆) 8 0.54 (s, 9 H), 0.49–0.68 (m, 2 H), 0.84–2.05 (m, 28 H), 2.96 (dd, 1 H), 3.64 (dd, 1 H), 3.76 (dd 1 H); ¹³C NMR (62.9 MHz, C₆D₆) δ 0.52, 14.22, 14.92, 24.56, 26.26, 26.34, 26.53, 26.71, 28.54, 29.28, 30.24, 30.82, 34.09, 39.70, 40.84, 42.31, 49.51, 82.89, 83.74, 111.65, 136.62, 188.68, 211.52; minor diastereomer:¹H NMR (250 MHz, C₆D₆) δ 0.51 (s, 9 H), 0.48-2.30 (m, 30 H), 2.71 (dd, 1 H), 3.52 (dd, 1 H), 3.79 (dd, 1 H).

15. The relative configuration of compounds **5a-c** at C-5 could not be established by 2D-NMR (NOESY) spectroscopy.

16. Cf. G. Snatzke, Angew. Chem. 1979, 91, 380-393; Angew. Chem. Int. Ed. Engl. 1979, 18, 363.

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